

Combinatorial Discovery of Peptide Dendrimer Enzyme Models Hydrolyzing Isobutyryl Fluorescein

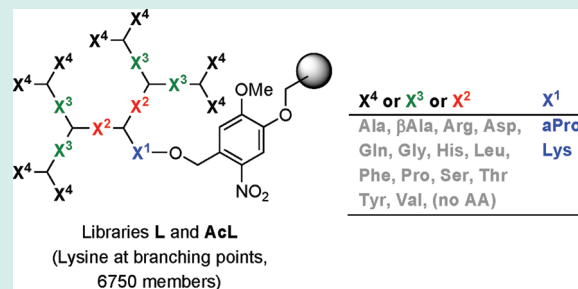
Noélie Maillard, Rasomoy Biswas, Tamis Darbre, and Jean-Louis Reymond*

Department of Chemistry and Biochemistry, University of Berne, Freiestrasse 3, CH-3012, Berne, Switzerland

S Supporting Information

ABSTRACT: Two 6750-membered one-bead-one-compound peptide dendrimer combinatorial libraries **L** ($(X^4)_8(LysX^3)_4(LysX^2)_2LysX^1$ (X^{1-4} = 14 different amino acids or deletion, *Lys* = branching lysine residue) and **AcL** (with N-terminal acetylation) were prepared by split-and-mix solid phase peptide synthesis. Screening toward fluorogenic substrates for esterase and aldolase activities using the *in silica* off-bead assay (N. Maillard et al., *J. Comb. Chem.* 2009, 11, 667–675) and bead decoding by amino acid analysis revealed histidine containing sequences active against fluorescein diacetate. Isobutyryl fluorescein, a related hydrophobic fluorogenic substrate, was preferentially hydrolyzed by dendrimers from library **AcL** containing hydrophobic residues such as **AcH3** (**AcHis**)₈(*LysLeu*)₄(*LysVal*)₂*LysLysOH*, compared to simple oligohistidine peptides as reference catalysts. Polycationic dendrimers from library **L** with multiple free N-termini such as **H8** (**His**)₈(*LysβAla*)₄(*LysThr*)₂*LysProNH*₂ (*aPro* = (2*S*,4*S*)-4-aminoproline) showed stronger reactivity toward 8-acetoxypyrene-1,3,6-trisulfonate with partial acylation of N-termini. These experiments highlight the role of noncatalytic amino acids to determine substrate selectivity in peptide dendrimer esterase models.

KEYWORDS: combinatorial libraries, peptide dendrimer, hydrolysis, isobutyryl fluorescein, enzyme models



INTRODUCTION

Dendrimers are regularly branched tree-like synthetic macromolecules displaying a range of interesting properties.¹ Most dendrimers consist of predefined organic cores to which variable end groups are attached for function.² Recently we reported that peptide dendrimers can be obtained by solid-phase peptide synthesis alternating branching diamino acids and standard α -amino acids in the sequence.³ Peptide dendrimers resemble proteins in their composition and adopt a globular shape because of their topology rather than by folding. In approaches similar to those reported by others for libraries of linear peptides,^{4,5} our screening of one-bead-one-compound (OBOC) combinatorial libraries⁶ identified peptide dendrimers with various functions such as esterase⁷ and aldolase⁸ catalysis and selective binding to metal ions,⁹ cofactors,¹⁰ and proteins.¹¹ In the case of enzyme model studies for esterase activities, our experiments best succeeded when using charged fluorogenic substrates such as 8-acetoxypyrene-1,3,6-trisulfonate **4** (Scheme 1). On the other hand screening with hydrophobic, non charged substrates generally failed to identify significant catalysis in any of the libraries.

The peptide dendrimer combinatorial libraries screened for esterase catalysis were based on sequence designs in which each amino acid appeared only twice in the synthesis in two successive branches. This design produced library sizes manageable by SPPS, which is limited by the number of solid support beads in the experiment, and enabled decoding by the inexpensive and efficient amino acid analysis (AAA) of the beads. However, with

this approach only a small fraction of the theoretically possible peptide dendrimer sequence space was actually surveyed. Indeed in a typical third generation peptide dendrimer library with eight variable amino acid positions with sequence $(X^8X^7)_8(BX^6X^5)_4(BX^4X^3)_2BX^2X^1$ (X^{1-8} = variable amino acid, *B* = branching diamino acid), a choice of 15 amino acids implied that over 2 billion possible sequences were possible, while we used only four different amino acids per variable position and thus prepared 65,536 sequences, corresponding to only 0.0026% of the theoretically possible sequence space. This left open the possibility that the difficulty to identify esterase catalysts toward hydrophobic substrates was caused by incomplete screening of sequence space.

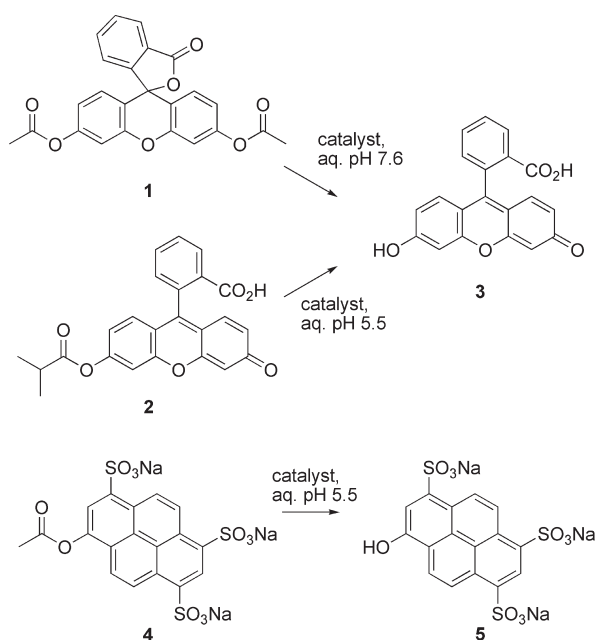
In our efforts to realize a more thorough coverage of peptide dendrimer sequence space, we herein report the screening of the 6750-membered combinatorial libraries **L** $(X^4)_8(LysX^3)_4(LysX^2)_2LysX^1$ (X^{1-4} = variable amino acid, *Lys* = branching lysine residue) and **AcL** (with N-terminal acetylation, Figure 1) with more extensive survey of sequence space for three of its four variable positions. The library was screened for conversion of various fluorogenic substrates for aldolase and esterase activities using our recently reported off-bead catalysis assay, which allows to assay the libraries in solution on the surface of the silica gel

Received: January 13, 2011

Revised: March 18, 2011

Published: March 25, 2011

Scheme 1. Peptide Dendrimer Catalyzed Fluorogenic Ester Hydrolysis Reactions



plate, a method which is less prone to bead-surface artifacts than simple on-bead activity assays.¹² The screening led to the identification of several sequences catalyzing the hydrolysis of fluorescein diacetate **1**. A comparative kinetic study with isobutyryl fluorescein **2** as a model hydrophobic substrate and 8-acetoxypyrene-1,3,6-trisulfonate **4** as a typical polyanionic substrate reveals unprecedented aspects of structure-activity relationships in peptide dendrimer enzyme models, in particular the role of noncatalytic amino acids to determine substrate selectivity.

RESULTS AND DISCUSSION

Library Design. AAA is applicable for decoding SPPS synthesis beads because the amount of peptide per bead is sufficient for the sensitivity of AAA, which requires at least 90 pmols. While our previous examples of AAA decodable libraries were limited to cases where each amino acid is only used in two of the variable positions during library synthesis, the method is theoretically suitable for decoding an exhaustive combinatorial library of third generation peptide dendrimers if each branch bears only a single variable amino acid position, as shown in Table 1.

Taking into account the possible difficulties in uniquely resolving the copy number above 10 copies by AAA, we set out to explore this exhaustive library design in a focused example considering the variable positions X^2 , X^3 , and X^4 (Figure 1). The core position X^1 would be limited to 4-aminoproline or lysine only to selectively explore single-site enamine-type organocatalytic aldolase dendrimers, a type of activity previously identified in peptides¹³ and peptide dendrimers.⁸ Fourteen amino acids were included for the variable positions including aromatic (phenylalanine and tyrosine), hydrophobic (leucine, proline, and valine), small and polar (alanine, glutamine, glycine, serine, and threonine), charged and catalytic residues (arginine, aspartate, histidine). β -alanine and a deletion (which can be identified

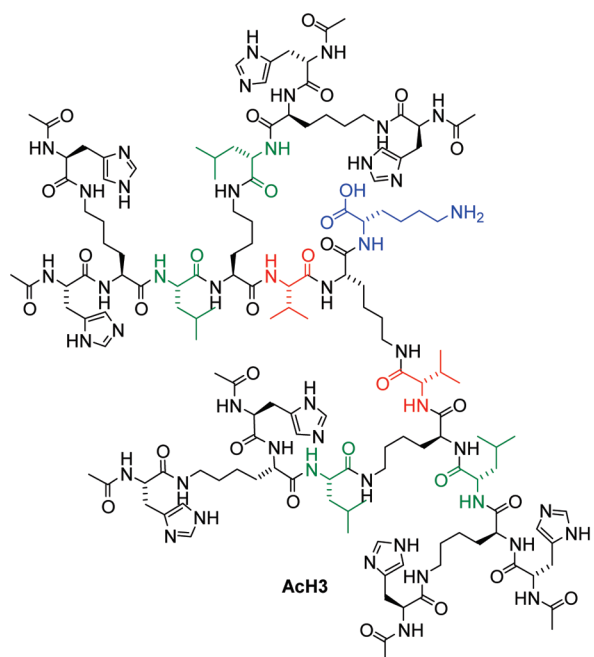
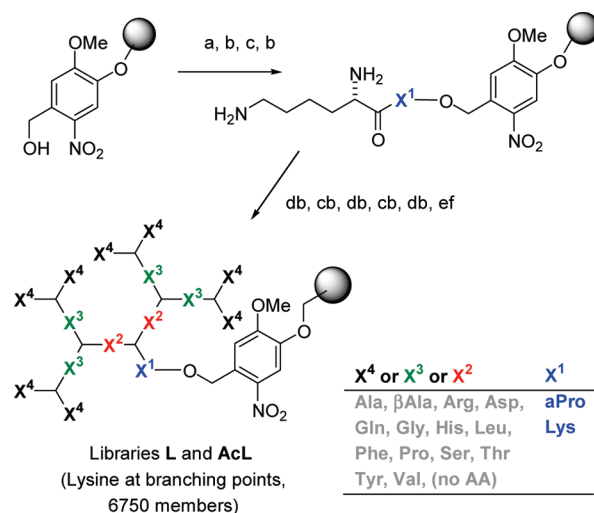


Figure 1. Synthesis of third generation peptide dendrimer combinatorial libraries AcL and L. Conditions: (a) MSNT (1-(2-mesitylenesulfonyl)-3-nitro-1H-1,2,4-triazole), Fmoc X^1 OH, 1-methylimidazole, CH_2Cl_2 , 1 h, ($2\times$); (b) piperidine 20% in DMF (2×10 min); (c) FmocLys(Fmoc)OH, PyBop, iPrNEt₂, NMP; (d) split in 15 portions, then Fmoc X^1 OH, PyBop, iPrNEt₂, NMP; (e) for AcL: $\text{Ac}_2\text{O}/\text{CH}_2\text{Cl}_2$ (1:1), 1 h; (f) TFA/TIS/ H_2O (94:5:1), 4 h. aPro = (2S,4S)-4-aminoproline, introduced with (*N*-Boc-*trans*-4-*N*-Fmoc-amino-L-proline).

by AAA) were also included to allow size variation in the dendrimer. We excluded tryptophan (not detectable by AAA), glutamate and asparagine (indistinguishable in AAA from glutamine respectively aspartate), isoleucine (too similar to leucine), lysine (already used as branching unit), and sulfur containing amino acids.

Library Synthesis and Screening. The library was prepared by split-and-mix Fmoc-SPPS on a 1.5 g batch of hydroxymethyl-photolinker NovaSyn TG resin (0.24 mmol/g, bead diameter 130 μm , $\sim 8\times 10^5$ bead/g, 200–300 pmol/bead). After the final Fmoc deprotection, library L was either used directly, thus

Table 1. Amino Acid Copy Numbers from AAA and Position Attribution in a 3rd Generation Peptide Dendrimer Library (X^4)₈(LysX³)₄(LysX²)₂LysX¹

copy number from AAA	positions occupied
1	X ¹
2	2 X ²
3	2 X ² + X ¹
4	4 X ³
5	4 X ³ + X ¹
6	4 X ³ + 2 X ²
7	4 X ³ + 2 X ² + X ¹
8	8 X ⁴
9	8 X ⁴ + X ¹
10	8 X ⁴ + 2 X ²
11	8 X ⁴ + 2 X ² + X ¹
12	8 X ⁴ + 4 X ³
13	8 X ⁴ + 4 X ³ + X ¹
14	8 X ⁴ + 4 X ³ + 2 X ²
15	8 X ⁴ + 4 X ³ + 2 X ² + X ¹

featuring multiple free N-termini with possible aldolase type catalytic properties, or acetylated to provide library **AcL** in which only the lysine or aminoproline residue at position X¹ displayed an enamine reactive amino acid. The library quality was checked by submitting 72 beads to AAA. A total of 56 beads (78%) returned a readable sequence, showing the expected even distribution of amino acids (Supporting Information, Table S1). The remaining beads contained 10 incomplete sequences (14%) and 6 empty beads (8%) reflecting the yield of library synthesis.

Screening was performed by off-bead screening following our recently reported protocol.¹² In this assay the library beads are partially photolyzed in the absence of solvent, and subsequently placed on the surface of a silicagel plate freshly impregnated with a solution of a fluorogenic substrate for the reaction of interest. The photolyzed portion of the dendrimer dissolves and diffuses into the surrounding wet silicagel. Synthesis beads carrying an active sequence are detected by the formation of a fluorescent halo indicating product formation. The beads are then retrieved with a pipet, washed, and subjected to decoding by AAA of the nonphotolyzed portion of the dendrimer which has remained attached to the bead. Decoding requires only total hydrolysis under acidic conditions, amino acid derivatization, and a single HPLC analysis per bead to give the amino acid composition. This method is much simpler than Edman sequencing, does not require a free N-terminus, and can even be adapted to the analysis of linear and cyclic peptide libraries.¹⁴

The procedure was applied by photolyzing 10 mg portions of libraries **L** and **AcL** in the absence of solvent and applying the beads to silica gel plates freshly impregnated with aqueous buffered solutions of the fluorogenic substrates at 200 μM in the case of the water-soluble substrates **6** and **8**. For the water-insoluble substrate **1** and **9** the silicagel plates were first impregnated with a 100 μM solution of the fluorogenic substrate in CH₂Cl₂, dried, and impregnated with aqueous buffer prior to applying the library beads. Substrates **6a/b**, **8**, and **9** release hydroxycoumarin by retroaldol reaction (**6a/b**), enediol formation, and tautomerization (**8**) or ester hydrolysis (**9**).

The tests with the aldol-type fluorogenic substrates **6**, **8**, and **9** were all negative, suggesting that enamine-type reactivity, which

Scheme 2. Reactions of Fluorogenic Substrates That Gave No Hits in the off-Bead Assays with Libraries **L and **AcL****

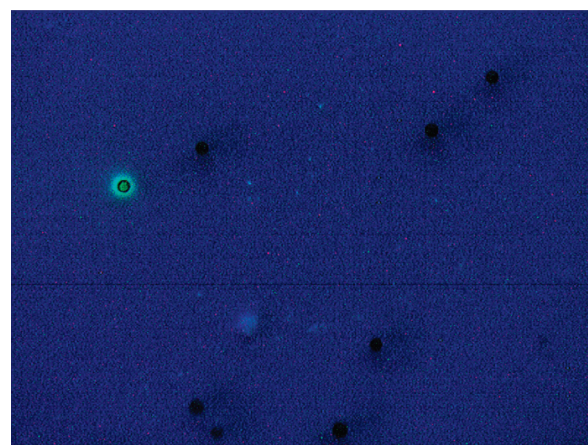
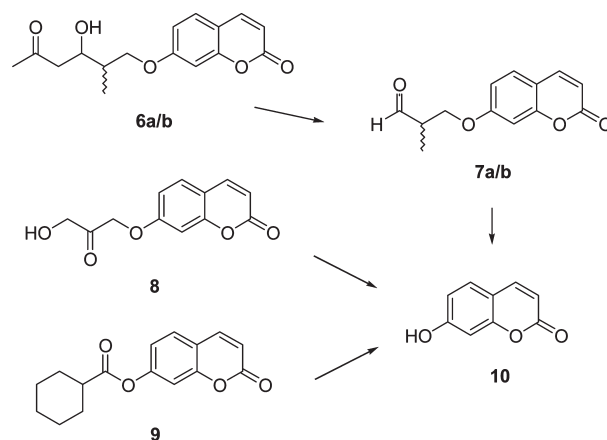


Figure 2. Fluorescent active halo from an off-bead screening with fluorescein diacetate **1**. The photolyzed beads (here **AcL**) were poured on an impregnated TLC with PB buffer 60 mM pH 7.6 and 100 μM **1**. The picture is taken under illumination with a 366 nm TLC-UV lamp.

was expected to involve with either the primary or the secondary α-amino group of the C-terminal amino acid at the core position X¹ in the case of library **AcL** or the multiple N-terminal amino group in the case of library **L**, did not occur at a detectable level in this assay (Scheme 2). While the hydrophobic esterase substrate **9** was also unreactive, active beads were observed in the case of the esterase substrate **1** for both libraries **L** and **AcL** (Figure 2). In each assay with **1** a fluorescent halo appeared after 24 h at room temperature (RT) around 7–9 beads (0.1% of approximately 8,000 beads). These active beads as well as a selection of control beads without halo were transferred to vials and subjected to AAA. From 50 beads picked in the assays with **AcL** and **L**, 37 returned a readable sequence (74%, Table 2), 8 contained incomplete sequences (16%), and 5 were empty (10%). The analysis of the 60 non-hits provided 43 readable sequences (72%) (Supporting Information, Table S2), 6 incomplete sequences (10%), and 11 empty beads (18%). It should be noted that approximately 5–10% of the positive hit beads were aggregated in groups of 2–5 beads, in which case the beads were not analyzed.

Table 2. Sequences for 37 Positive Hits in the off-Bead Assay with Fluorescein Diacetate 1

library	no.	X ⁴	X ³	X ²	X ¹	
AcL	1	His	Leu	Val	Lys	
	2	AcH1	His	Phe	Arg	Lys
	3		His	Gln	Ala	Lys
	4		AcH2	His	Gln	His
	5	His		Val	Leu	Lys
	6	His	Tyr	Arg	Lys	
	7	His	β Ala	Ser	aPro	
	8	AcH3	His	Leu	Val	Lys
	9		His	Val	His	aPro
	10	His	β Ala	Thr	aPro	
	11	AcH4	His	β Ala	Arg	aPro
	12		His	Gly	Val	aPro
	13	Ser	Arg	His	aPro	
	14	His	Ala	Ala	aPro	
	15	AcH5	Gly	His	His	aPro
	16		His	His	aPro	
	17		Arg	Leu	His	aPro
L	18	His	Leu	Leu	Lys	
	19	Phe	His	Arg	aPro	
	20	Gln	Phe	His	aPro	
	21	H6	His	Leu	Ala	Lys
	22		H7	His	Phe	Pro
	23	Phe	His	Phe	aPro	
	24	H8	His	bAla	Thr	aPro
	25		Thr	His	His	aPro
	26	H9	His		Gly	Lys
	27		Pro	His	Arg	Lys
	28	H10	Phe	His	His	Lys
	29		His	Phe	Val	Lys
	30	Phe	His	Ala	aPro	
	31	His	Leu	Ser	aPro	
	32	Leu	His	His	aPro	
	33	His		Ala	aPro	
	34	His	Leu	Pro	aPro	
	35	Phe	His	Ser	aPro	
	36	His	Arg	Arg	aPro	
	37	His	Phe	Ala	Lys	

Control and inactive beads showed approximately random amino acid composition. By contrast the sequences on active beads showed a strong consensus for histidine (Figure 3). All positive hits in libraries AcL and L had at least one histidine in the sequence, most frequently at the outer position X⁴ (8 copies), such that the average amino acid composition comprised approximately 50% histidine. Histidine at position X⁴ was generally combined with hydrophobic amino acids at the inner positions X³ and X², in particular phenylalanine and leucine. The cationic arginine occurred in 9 of the 37 positive hits, but the anionic aspartate was totally absent from positive hits.

Hit Resynthesis and Kinetic Studies. Ten hits and four non-hits from screening with fluorescein diacetate 1 were selected equally from libraries AcL and L and resynthesized by SPPS (Table 3), using either Wang or Rink amide resin yielding a

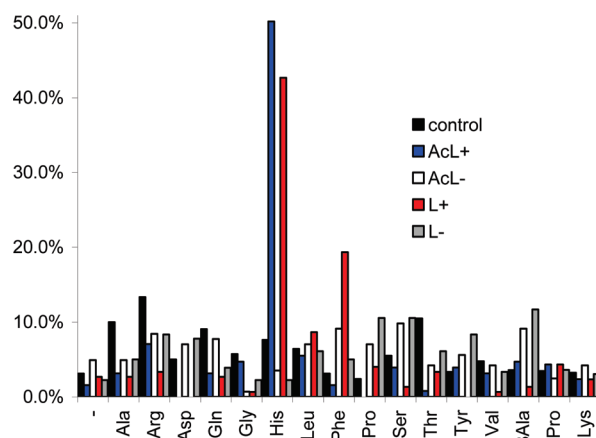


Figure 3. Composition of the dendrimer sequences. **Control:** beads picked randomly from library L (56 beads). **AcL+:** positive hits from AcL with fluorescein diacetate 1 (17 beads); **AcL-:** inactive beads from AcL with 1 (19 beads); **L+:** positive hits from L with 1 (20 beads); **L-:** inactive hits from L with 1 (24 beads). The total number of occurrences of each amino acid is calculated as $8m + 4n + 2p + q$ with m , n , p , and q denoting the number of sequences with that amino acid at positions X⁴, X³, X², and X¹, respectively.

Table 3. Synthesis of Hits and Non-Hit Peptide Dendrimers Identified by off-Bead Screening with Fluorescein Diacetate 1 with OBOC Libraries AcL and L

no.	sequence	yield, mg (%)	MS calc/obs	MS obsd
AcH1	(AcH) ₈ (KF) ₄ (KR) ₂ KKO ₂ H	66.2 mg (18%)	3376.8	3377
AcH2	(AcH) ₈ (KQ) ₄ (KH) ₂ KaPNH ₂	21.5 mg (8%)	3245.7	3246
AcH3	(AcH) ₈ (KL) ₄ (KV) ₂ KKO ₂ H	50.3 mg (16%)	3126.8	3127
AcH4	(AcH) ₈ (K β A) ₄ (KR) ₂ KaPNH ₂	2.4 mg (1%)	3055.7	3056
AcH5	(AcG) ₈ (KH) ₄ (KH) ₂ KaPNH ₂	2.0 mg (1%)	2641.4	2642
H6	(H) ₈ (KL) ₄ (KA) ₂ KKO ₂ H	62.0 mg (22%)	2734.7	2735
H7	(H) ₈ (KF) ₄ (KP) ₂ KaPNH ₂	15.8 mg (7%)	2905.6	2906
H8	(H) ₈ (K β A) ₄ (KT) ₂ KaPNH ₂	3.3 mg (2%)	2609.5	2609
H9	(H) ₈ (K) ₄ (KG) ₂ KKO ₂ H	73.8 mg (29%)	2254.3	2254
H10	(F) ₈ (KH) ₄ (KH) ₂ KKO ₂ H	64.0 mg (21%)	3042.7	3043
AcN1	(AcS) ₈ (K) ₄ (KY) ₂ KaPNH ₂	7.9 mg (5%)	2385.2	2385
AcN2	(AcF) ₈ (KR) ₄ (KH) ₂ KKO ₂ H	28.8 mg (9%)	3454.9	3455
N3	(A) ₈ (KP) ₄ (KQ) ₂ KaPNH ₂	1.3 mg (1%)	2239.4	2239
N4	(S) ₈ (KV) ₄ (KH) ₂ KKO ₂ H	46.5 mg (22%)	2410.4	2410

C-terminal carboxyl or carboxamide (the nature of the C-terminus turned out to have negligible influence on catalysis). The dendrimers were obtained in good yields and purity after purification by RP-HPLC. The hit sequences carried 6 to 10 histidine residues. Two of the non-hits had no histidine residues and two had a pair of histidines at position X². A series of linear oligohistidine peptides from 1 to 15 residues (**His1-His15**) and the histidine-lysine oligomer **P25** (AcHKHKHKHKHKHNH₂), which were known to exhibit good esterolytic activity toward **4**,¹⁵ were also included in the study. Kinetics were measured in 96-well microtiter plates following product formation by fluorescence in aqueous buffered solutions containing the peptide dendrimer at 5 μ M and the substrates **1**, **2** or **4**, including 17% acetonitrile cosolvent with **1** and 11% acetonitrile cosolvent with **2** (Table 4 and Supporting Information, Table S3, Figure 4).

Table 4. Kinetic Parameters for Dendrimer and Peptide Catalyzed Ester Hydrolysis

no.	sequence	N_{His}	$k_{\text{cat}}/K_{\text{M}}$ (1) ^a $\text{M}^{-1} \text{min}^{-1}$	$k_{\text{cat}}/K_{\text{M}}$ (2) ^b $\text{M}^{-1} \text{min}^{-1}$	$k_{\text{cat}}/K_{\text{M}}$ (4) ^c $\text{M}^{-1} \text{min}^{-1}$
AcH1	(AcH) ₈ (KF) ₄ (KR) ₂ KCOH	8	24 ± 2	112 ± 7	745 ± 30
AcH2	(AcH) ₈ (KQ) ₄ (KH) ₂ KaPNH ₂	10	34 ± 8	160 ± 23	1100 ± 47
AcH3	(AcH) ₈ (KL) ₄ (KV) ₂ KCOH	8	25 ± 9	260 ± 70	910 ± 30
AcH4	(AcH) ₈ (KβA) ₄ (KR) ₂ KaPNH ₂	8	19 ± 0.6	170 ± 20	1170 ± 230
AcH5	(AcG) ₈ (KH) ₄ (KH) ₂ KaPNH ₂	6		47 ± 5	1570 ± 110
H6	(H) ₈ (KL) ₄ (KA) ₂ KCOH	8	20 ± 5	33 ± 8	1770 ± 65
H7	(H) ₈ (KF) ₄ (KP) ₂ KaPNH ₂	8	47 ± 7	310 ± 70	3820 ± 260
H8	(H) ₈ (KβA) ₄ (KT) ₂ KaPNH ₂	8	35 ± 12	36 ± 8	3400 ± 80
H9	(H) ₈ (K) ₄ (KG) ₂ KCOH	8	21 ± 10	30 ± 3	2770 ± 26
H10	(F) ₈ (KH) ₄ (KH) ₂ KCOH	6	43 ± 19	130 ± 6	1690 ± 40
AcN1	(AcS) ₈ (K) ₄ (KY) ₂ KaPNH ₂				
AcN2	(AcF) ₈ (KR) ₄ (KH) ₂ KCOH	2		68 ± 9	670 ± 33
N3	(A) ₈ (KP) ₄ (KQ) ₂ KaPNH ₂				137 ± 12
N4	(S) ₈ (KV) ₄ (KH) ₂ KCOH	2		18 ± 5	660 ± 13
His1	AcHNH ₂	1	nd	5 ± 0.5	3 ± 0.2
His2	AcHHNH ₂	2	nd	10 ± 1	27 ± 0.4
His3	AcHHHNH ₂	3	nd	19 ± 1	132 ± 4
His4	AcHHHHNH ₂	4	nd	25 ± 2	220 ± 6
His5	AcHHHHHNH ₂	5	42 ± 14	36 ± 5	270 ± 7
His6	AcHHHHHNH ₂	6	39 ± 12	35 ± 3	380 ± 2
His7	AcHHHHHNH ₂	7	36 ± 13	42 ± 3	796 ± 130
His8	AcHHHHHNH ₂	8	43 ± 16	60 ± 7	960 ± 140
His9	AcHHHHHNH ₂	9	41 ± 20	65 ± 8	1010 ± 130
His10	AcHHHHHNH ₂	10	56 ± 27	81 ± 5	1240 ± 110
His11	AcHHHHHNH ₂	11	49 ± 32	59 ± 2	1200 ± 130
His12	AcHHHHHNH ₂	12	nd	70 ± 7	1540 ± 180
His13	AcHHHHHNH ₂	13	77 ± 46	75 ± 4	1570 ± 250
His14	AcHHHHHNH ₂	14	83 ± 41	79 ± 3	1690 ± 220
His15	AcHHHHHNH ₂	15	nd	114 ± 13	2210 ± 190
P25	AcHKHKHKHKHNH ₂	6	nd	33 ± 3	2380 ± 200

^a Conditions: 3.75–15 μM peptide or dendrimer, 3–100 μM **1**, 20 mM phosphate pH 7.6 with 17% acetonitrile, 34 °C. ^b Conditions: 3.75–20 μM peptide or dendrimer, 3–100 μM **2**, 5 mM citrate pH 5.5, with 11% acetonitrile, 34 °C. ^c Conditions: 3.75–20 μM peptide or dendrimer, 58.5–1000 μM substrate **4**, 5 mM citrate pH 5.5, 34 °C. Spontaneous background reaction $k_{\text{uncat}} = 3.22 \times 10^{-4} \text{ min}^{-1}$ for **1**, $3.37 \times 10^{-5} \text{ min}^{-1}$ for **2**, $3.58 \times 10^{-5} \text{ min}^{-1}$ for **4**. k_2 is the 2nd order rate constant with 4-methyl imidazole, $k_2 = 2.26 \text{ M}^{-1} \text{ min}^{-1}$ for **2**, $k_2 = 1.0 \text{ M}^{-1} \text{ min}^{-1}$ for **4**. 120 μL assays in microtiterplate wells were followed by fluorescence ($\lambda_{\text{exc}} = 450 \pm 25 \text{ nm}$, $\lambda_{\text{em}} = 530 \pm 12 \text{ nm}$). All kinetics were run in triplicate. See Experimental Section and Supporting Information, Table S3 for complete data.

The hydrolysis of fluorescein diacetate **1** to fluorescein was investigated in phosphate buffer pH 7.6 in the presence of 17% acetonitrile cosolvent to ensure substrate solubility. Unfortunately experiments showed poor reproducibility and only provided an estimation of the apparent specificity constant $k_{\text{cat}}/K_{\text{M}}$. The data confirmed the screening results with all but one positive hit showing the expected activity and the four non-hit sequences showing no activity with this substrate. The apparent activity was comparable to that observed with linear oligohistidine peptides when counted per catalytic histidine residues.

To better characterize the unprecedented esterolytic activity of dendrimers with hydrophobic substrates, we prepared isobutyryl fluorescein **2** (Scheme 1). This substrate requires only a single ester bond cleavage to release the fluorescent product fluorescein, and was expected to show higher aqueous solubility because of the presence of a free carboxyl group. We also anticipated a better reproducibility of the kinetic data from our previous experiences with related straight-chain aliphatic fluorescein monoesters as lipase and esterase substrates.¹⁶ Indeed

substrate **2** gave better reproducible data than **1**, and showed activity both at pH 7.6 and at pH 5.5. The best kinetic data were collected at pH 5.5; however, the substrate concentration was held at 100 μM or below because of limited solubility, which precluded a determination of the Michaelis–Menten constant K_{M} . The data were therefore interpreted in terms of the specificity constant $k_{\text{cat}}/K_{\text{M}}$. The kinetic study was also carried out with the trianionic substrate **4** at the same pH because of its strong reactivity with histidine containing catalysts.

Structure-Activity Relationships. A comparative analysis of the catalytic proficiencies of the different peptide dendrimers with the hydrophobic ester substrate **2** and the polyanionic substrate **4** provided a useful insight into the structure-activity relationships. For each substrate the relative specificity constants compared to the reference small molecule catalyst 4-methylimidazole were computed as $(k_{\text{cat}}/K_{\text{M}})/k_2(\mathbf{2})$ and $(k_{\text{cat}}/K_{\text{M}})/k_2(\mathbf{4})$, respectively $((k_{\text{cat}}/K_{\text{M}})/k_2)/N_{\text{His}}(\mathbf{2})$ and $((k_{\text{cat}}/K_{\text{M}})/k_2)/N_{\text{His}}(\mathbf{4})$ for the effect counted per histidine residue (Figure 5A and 5B).

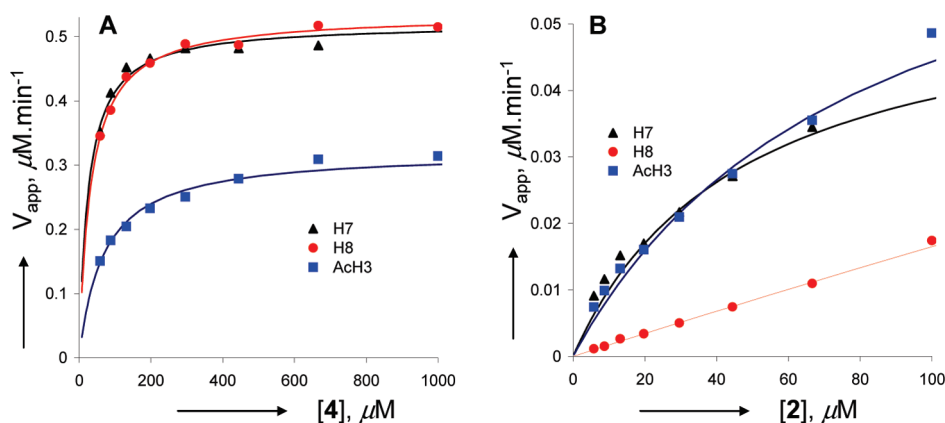


Figure 4. Catalytic reaction rates with peptide dendrimers H7, H8, and Ach3. Conditions: aq. 5 mM citrate pH 5.5 with 5 μM dendrimer.

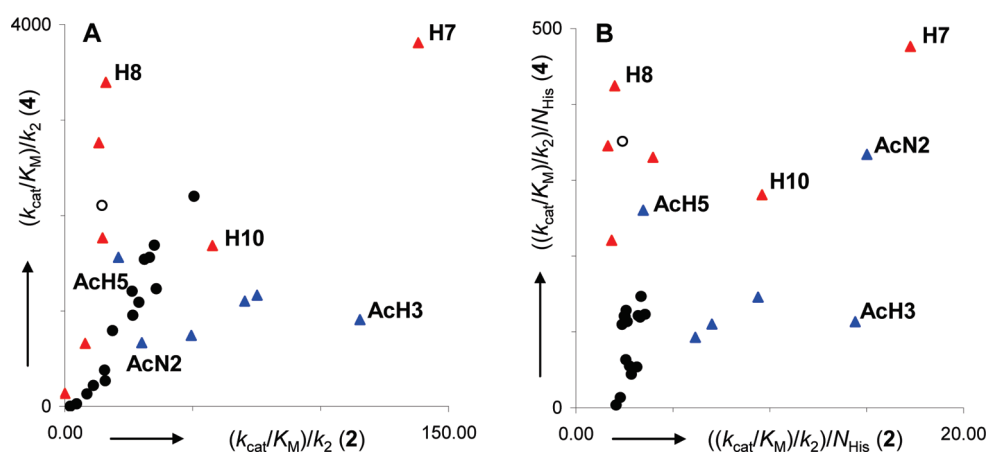


Figure 5. Relative catalytic proficiencies of histidine containing peptides and peptide dendrimers with substrates 2 and 4 in aqueous 5 mM citrate buffer pH 5.5. (A) Relative catalytic proficiencies $(k_{\text{cat}}/K_{\text{M}})/k_2$, k_2 is the 2nd order rate constant with 4-methyl-imidazole as reference catalyst. (B) Relative catalytic proficiencies per histidine residue $((k_{\text{cat}}/K_{\text{M}})/k_2)/N_{\text{His}}$, N_{His} is the number of histidine residue in the peptide or peptide dendrimer catalyst. (red solid triangles): peptide dendrimers from library L; (blue solid triangles): peptide dendrimers from library AcL; (black solid circles): oligohistidine peptide from 1 to 15 amino acid residues, with N-terminal acetylation. (black open circles): octapeptide AcHKHKHKHKNH₂. See Table 4 for detailed kinetic data and conditions.

The reference oligohistidine peptides His1-His15 showed increasing relative catalytic proficiencies $(k_{\text{cat}}/K_{\text{M}})/k_2$ as a function of length for both substrates. The relative catalytic proficiencies for substrate 4 were on average 40-fold larger than for substrate 2. The catalytic effect per histidine residue for the trianionic substrate 4 $((k_{\text{cat}}/K_{\text{M}})/k_2)/N_{\text{His}}$ (4) strongly increased up to seven histidine residues before leveling off. By contrast, the catalytic proficiency per histidine residue for the isobutryl fluorescein substrate 2 $((k_{\text{cat}}/K_{\text{M}})/k_2)/N_{\text{His}}$ (2) was essentially independent of peptide length. Both the higher catalytic activity with 4 compared to 2 and the length dependence of $((k_{\text{cat}}/K_{\text{M}})/k_2)/N_{\text{His}}$ (4) may be explained by significant electrostatic binding with substrate 4 requiring cooperative action of several histidine residues, while the more hydrophobic substrate 2 does not bind and is therefore insensitive to peptide length.

The peptide dendrimers showed a differentiated catalytic behavior toward 2 and 4, with several dendrimers standing out by their reactivities, in particular relative to the linear peptides. Dendrimers with acetylated N-termini from library AcL were generally more active toward the hydrophobic fluorescein

substrate 2. The most selective catalyst for substrate 2 was peptide dendrimer Ach3 (AcHis)₈(LysLeu)₄(LysVal)₂LysLysOH, which combined eight histidine residues in the third generation branches with hydrophobic residues in the first and second generation branches. On the other hand peptide dendrimer Ach5 (AcGly)₈(LysHis)₄(LysHis)₂LysaProNH₂ with six histidine residues in the inner first and second generation branches and only glycine residue in the third generation branch was more selective toward the trianionic substrate 4. Interestingly the negative control dendrimer AcN2 (AcPhe)₈(LysArg)₄(LysHis)₂LysLysOH, which was not active with fluorescein diacetate 1, showed significant activity with the isobutrylate 2 as well as with the trisulfonate 4. This reactivity stood out in particular when considering the specific catalytic effect per histidine residue, and probably reflects substrate binding interactions with the arginine side chains.

The peptide dendrimers from library L with free N-termini were more reactive toward the trianionic substrate 4, in particular for dendrimer H8 (His)₈(LysβAla)₄(LysThr)₂LysaProNH₂, reflecting electrostatic substrate binding. This dendrimer showed only poor reactivity with 2 attributable to the absence of

hydrophobic residues. Peptide dendrimer **H7** (His)₈(LysPhe)₄(LysPro)₂LysaProNH₂ on the other hand had the strongest reactivity of all catalysts toward both substrates. A particularly favorable combination of eight catalytic histidine residues at the N-terminus with multiple positive charges from the free N-termini favored activity with **4**. In addition the hydrophobic interior of **H7** constituted of four phenylalanines in the second generation branch and two proline residues in the first generation branch favored activity with **2**. Peptide dendrimer **H10** (Phe)₈(LysHis)₄(LysHis)₂LysLysOH featuring an inverted combination with eight hydrophobic phenylalanines in the third generation branches and six catalytic histidines in the inner branches showed a weaker but similarly balanced reactivity between **2** and **4**. By comparison the linear peptide **P25** AcHKHKHKHKHKHNH₂ which combines six histidine residues with five ammonium group from lysine side chains showed only enhanced catalysis with the trisulfonate **4** but not with isobutyryl fluorescein **2**, because of the absence of hydrophobic residues.

Mass Spectrometric Analysis. As previously observed with **P25**, the peptide dendrimers from library **L** underwent significant acylation of the N-termini in contact with substrate **4**, as detected by MS for the reaction of **H9** and **H10** under high substrate and dendrimer concentration (see Supporting Information). In the case of dendrimer **H9** MS analysis also showed the formation of substrate-dendrimer complexes with one, two, and three substrates per dendrimer, in agreement with the multivalent nature of these catalysts. On the other hand acylation with isobutyrate **2** occurred only to a very low level, probably because of the slower reaction rates and the bulkier acyl group in that substrate.

CONCLUSION

The experiments above showed the first example of a combinatorial peptide dendrimer library extensively covering its sequence space by using 14 amino acids and a deletion at each variable position, exploiting the fact that such exhaustive sampling is compatible with AAA for bead decoding. Screening of catalysis with fluorogenic substrates in the off-bead *in silica* assay uncovered specific reactivities with fluorescein diacetate **1**. While this substrate proved unsuitable for kinetic studies, the related isobutyryl fluorescein **2**, which requires only a single ester bond cleavage to release its fluorescent product, allowed the characterization of this reactivity. Structure-activity relationship studies comparing the catalyzed ester hydrolysis reaction rates of **2** with those of the trianionic substrate **4** showed that efficient hydrolysis of the hydrophobic substrate **2** requires a combination of multiple catalytic histidine residues with hydrophobic amino acids such as phenylalanine and leucine. These hydrophobic residues presumably enable substrate binding, although pre-equilibrium binding was not visible in the accessible concentration range. By contrast peptide dendrimers with mostly cationic residues and no hydrophobic groups showed stronger activity with the trisulfonate **4**. These experiments highlight the power of combinatorial approaches to identify peptidic enzyme models and the role of noncatalytic amino acids to determine substrate selectivity in these systems.

EXPERIMENTAL SECTION

General Procedures. All reagents were either purchased from Aldrich, Fluka, or Acors Organics. PyBOP, amino acids, and their

derivatives were purchased from Advanced ChemTech (U.S.A.) or Novabiochem (Switzerland). Amino acids were used as the following derivatives: Fmoc-Ala-OH, Fmoc-βAla-OH, Fmoc-Arg(Pbf)-OH, Fmoc-Asp(OtBu)-OH, Fmoc-Gln(Trt)-OH, Fmoc-Gly-OH, Fmoc-His(Boc)-OH, Fmoc-Ile-OH, Fmoc-Leu-OH, Fmoc-Lys(Boc)-OH, Fmoc-Lys(Fmoc)-OH, Fmoc-Phe-OH, Fmoc-Pro-OH, Fmoc-Ser(tBu)-OH, Fmoc-Thr(tBu)-OH, Fmoc-Tyr(tBu)-OH, Fmoc-Val-OH. Fmoc-^αPro(Boc)-OH was purchased from Aldrich. TG S RAM (loading: 0.23 or 0.24 mmol·g⁻¹) resin was purchased from Rapp Polymere (Germany) while hydroxymethyl-photolinker NovaSyn TG (loading: 0.24 mmol·g⁻¹) and TGA (loading: 0.26 mmol·g⁻¹) were purchased from Novabiochem. Peptide dendrimer syntheses were performed manually in polypropylene syringes fitted with a polyethylene frit, a Teflon stopcock and stopper. Analytical RP-HPLC was performed in Waters (996 Photo diode array detector) chromatography system using Atlantis column (dC18, 5 μm, 4.6 × 100 mm, flow rate 1.4 mL·min⁻¹). Analytical RP-UPLC was performed by Ultimate 3000 Rapid Separation LC System using diode array detector (DAD-3000RS) and Acclaim RSLC 120 C18 column (2.2 μm, 120 Å, 3.0 × 50 mm, flow 1.2 mL min⁻¹) from Dionex. Preparative RP-HPLC was performed in Waters Prep LC4000 chromatography system using a Delta-Pak column (C18, 15 μm, pore size 300 Å, flow rate 80 mL·min⁻¹) or an Atlantis OBD column (dC18, 5 μm, 19 × 100 mm, flow rate 20 mL·min⁻¹). Compounds were detected by UV absorption at 214 nm. All RP-HPLC were using HPLC-grade acetonitrile and mQ-deionized water. The elution solutions were A H₂O with 0.1% TFA; B H₂O/MeCN (50:50); C H₂O/MeCN (10:90) with 0.1% TFA; D H₂O/MeCN (40:60) with 0.1% TFA. MS spectra and amino acid analyses measurements were provided by Mass Spectrometry, and Protein Analysis services respectively of the Department of Chemistry and Biochemistry at the University of Bern. Kinetic measurements were carried out using a CytoFluor Series 4000 multiwell plate reader from PerSeptive Biosystems.

Synthesis of Split-and-Mix Library. The peptide dendrimer library was prepared from 1.5 g–100 mg per split-resin batch of hydroxymethyl-photolinker NovaSyn TG resin (0.24 mmol·g⁻¹) divided equally in 15 reactors. The attachment of the first amino acid to hydroxymethyl resin was done by using MSNT with 1-MeIm.¹⁷ The resin was placed in a syringe and swelled with dry CH₂Cl₂ under nitrogen. In a round-bottom flask, the appropriate amino acid (5 equiv) was dissolved in dry CH₂Cl₂ (3 mL per mmol) and a few drops of tetrahydrofuran (THF). MSNT (5 equiv) and 1-methylimidazole (3.75 equiv) were added, dissolved, and the mixture was added to the syringe. Then the syringe was agitated at RT for 1 h. The resin was washed with CH₂Cl₂ (5 × 5 mL). The coupling was repeated two times. A sensitive color test (alizarin-cyanuric chloride test)¹⁸ was used to detect the presence of hydroxyl groups on solid support; the presence of unreacted hydroxyl groups was indicated by red-colored beads. The Fmoc-protecting groups of the resin were removed with a solution of 20% piperidine in dimethylformamide (DMF, 2 × 10 min). For further couplings, the resin was acylated with one of the amino acid (3 eq/G, G = generation) in the presence of PyBOP (3 eq/G) and DIEA (5 eq/G) in NMP.^{3c} Amino acids, derivatives or diamino acids were coupled for 30 min (G0), 1 h (G1), 2 h (G2), 4 h (G3). The completion of the reaction was checked using 2,4,6-trinitrobenzenesulfonic acid solution (TNBS) or chloranil test.¹⁹ If the beads were red (brown for proline), there were some free amino

groups and the resin test was positive. If they were colorless, there were no more free amino groups, and the resin test was negative. The coupling was repeated after a positive test. After each coupling, the resin in each syringe was deprotected (20% piperidine in DMF, 2×10 min) followed by TNBS or chloranil test (test must be positive). Then the resin batches were mixed together, vortexed during 1 min, and split equally. These split-and-mix steps were repeated after each amino acid coupling. After each coupling or deprotection, the resin was successively washed with NMP, MeOH, and CH_2Cl_2 (3×5 mL, with each solvent). At the end of the synthesis, the Fmoc protected library was dried and stored at -18 °C. For all the steps, the syringes were covered with aluminum to prevent exposure to light.

Off-Bead Screening. Before the screening of the library, the Fmoc protecting groups were removed and the last amino acid was acetylated with $\text{Ac}_2\text{O}/\text{CH}_2\text{Cl}_2$ (1:1) for 1 h (library **AcL**) or not acetylated (library **L**). The side-chain protecting groups were removed with TFA/TIS (triisopropylsilane)/ H_2O (94:5:1) during 4 h, resulting in a functional dendrimer library on-beads. Screening was performed by washing the beads (5×3 mL) with the buffer and swelling those (10 mg, 2×1 h) in the same aqueous buffered solution. After solution filtration and drying under vacuum, the resin was poured on a Petri dish (diameter: 3 cm) and was partially irradiated under an Hg lamp (100 W) with a 366 nm filter for 30 min. A glass TLC plate (10×10 cm, SIL G25 without fluorescent indicator) was impregnated with 100 μM fluorescein diacetate **1** dissolved in CH_2Cl_2 , and dried. The TLC was sprayed with PB buffer (60 mM, pH 7.6) and the photolyzed beads were poured on the slightly wet TLC and incubated in a close Petri dish for 12 h. Under UV irradiation (366 nm) a green fluorescent halo appeared on the TLC around some of the beads. These active beads were picked and subjected to AAA.

Sequence Determination. Single dendrimer-containing resin beads were hydrolyzed with 6 M aqueous HCl at 110 °C for 22 h. The amino acids were derivatized with phenylisothiocyanate (PITC) and the phenylthiocarbonyl (PTC) derivatives analyzed on a RP-C18 Novapak column.

Dendrimer Synthesis. The resin (TG S RAM, or TGA) was swelled in CH_2Cl_2 , and the Fmoc-protecting groups of the resin were removed with a solution of 20% piperidine in DMF (2×10 min). Then the resin was acylated with each amino acid, derivative, or diamino acid (3 eq/G) using PyBOP (3 eq/G) and DIEA (5 eq/G) in NMP. The first coupling on the TGA resin used the described protocol (MSNT, 1-Melm). The completion of the reaction was checked using the TNBS or chloranil test (alizarin-cyanuric chloride test for the first coupling on TGA). At the end of the synthesis, the resin was acetylated with $\text{Ac}_2\text{O}/\text{CH}_2\text{Cl}_2$ (1:1) for 1 h for library **AcL**. The cleavage was carried out with TFA/TIS/ H_2O (94:5:1) during 4 h. The peptide was precipitated with *tert*-butylmethylether (MTBE) then dissolved in a $\text{H}_2\text{O}/\text{MeCN}$ mixture. Peptides were purified by preparative RP-HPLC and obtained as TFA salts after lyophilization. Unless otherwise mentioned, the gradient used for analytical HPLC is A/D = 100/0 to 0/100 in 10 min, 100 D in 5 min, $1.4 \text{ mL} \cdot \text{min}^{-1}$.

(*AchHis*)₈(*LysPhe*)₄(*LysArg*)₂*LysLysOH* (**Ach1**). Starting with 300 mg of TGA ($0.26 \text{ mmol} \cdot \text{g}^{-1}$), dendrimer **Ach1** was obtained as a white foamy solid after cleavage from the resin and preparative RP-HPLC purification (66.2 mg, 18%). Anal. RP-HPLC: $t_{\text{R}} = 8.1$ min. MS (ES+) calcd for $\text{C}_{160}\text{H}_{231}\text{N}_{52}\text{O}_{31}$ $[\text{M}+\text{H}]^+$: 3376.8, found: 3377.0; $[\text{M}+\text{K}]^+$: 3414.8, found: 3416.0; $[\text{M}+\text{TFA}]^+$: 3488.8, found: 3491.0.

(*AchHis*)₈(*LysGln*)₄(*LysHis*)₂*Lys^oProNH₂* (**Ach2**). Starting with 250 mg of TG S RAM ($0.24 \text{ mmol} \cdot \text{g}^{-1}$), dendrimer **Ach2** was obtained as a white foamy solid after cleavage from the resin and preparative RP-HPLC purification (21.5 mg, 8%). Anal. RP-HPLC: $t_{\text{R}} = 7.0$ min. MS (ES+) calcd for $\text{C}_{143}\text{H}_{214}\text{N}_{55}\text{O}_{34}$ $[\text{M}+\text{H}]^+$: 3245.7, found: 3246.0; $[\text{M}+\text{K}]^+$: 3287.7, found: 3285.0.

(*AchHis*)₈(*LysLeu*)₄(*LysVal*)₂*LysLysOH* (**Ach3**). Starting with 300 mg of TGA ($0.26 \text{ mmol} \cdot \text{g}^{-1}$), dendrimer **Ach3** was obtained as a white foamy solid after cleavage from the resin and preparative RP-HPLC purification (50.3 mg, 16%). Anal. RP-HPLC: $t_{\text{R}} = 8.3$ min. MS (ES+) calcd for $\text{C}_{146}\text{H}_{233}\text{N}_{46}\text{O}_{31}$ $[\text{M}+\text{H}]^+$: 3126.8, found: 3127.0; $[\text{M}+\text{Na}]^+$: 3149.8, found: 3149.0; $[\text{M}+\text{K}]^+$: 3165.8, found: 3167.0.

(*AchHis*)₈(*Lys β Ala*)₄(*LysArg*)₂*Lys^oProNH₂* (**Ach4**). Starting with 250 mg of TG S RAM ($0.24 \text{ mmol} \cdot \text{g}^{-1}$), dendrimer **Ach4** was obtained as a white foamy solid after cleavage from the resin and preparative RP-HPLC purification (2.4 mg, 1%). Anal. RP-HPLC: $t_{\text{R}} = 6.8$ min. MS (ES+) calcd for $\text{C}_{135}\text{H}_{212}\text{N}_{53}\text{O}_{30}$ $[\text{M}+\text{H}]^+$: 3055.7, found: 3056.0; $[\text{M}+\text{K}]^+$: 3093.7, found: 3097.0.

(*AcGly*)₈(*LysHis*)₄(*LysHis*)₂*Lys^oProNH₂* (**Ach5**). Starting with 250 mg of TG S RAM ($0.24 \text{ mmol} \cdot \text{g}^{-1}$), dendrimer **Ach5** was obtained as a white foamy solid after cleavage from the resin and preparative RP-HPLC purification (2.0 mg, 1%). Anal. RP-HPLC: $t_{\text{R}} = 7.0$ min. MS (ES+) calcd for $\text{C}_{115}\text{H}_{178}\text{N}_{43}\text{O}_{30}$ $[\text{M}+\text{H}]^+$: 2641.4, found: 2642.0; $[\text{M}+\text{K}]^+$: 2679.4, found: 2682.0; $[\text{M}+\text{TFA}]^+$: 2753.4, found: 2756.0.

(*His*)₈(*LysLeu*)₄(*LysAla*)₂*LysLysOH* (**H6**). Starting with 300 mg of TGA ($0.26 \text{ mmol} \cdot \text{g}^{-1}$), dendrimer **H6** was obtained as a white foamy solid after cleavage from the resin and preparative RP-HPLC purification (62.0 mg, 22%). Anal. RP-UPLC: $t_{\text{R}} = 1.312$ min (A/D = 100/0 to 0/100 in 2.2 min, 100 D in 1.3 min, A/D = 0/100 to 100/0 in 0.5 min, 100 A in 1 min $1.2 \text{ mL} \cdot \text{min}^{-1}$). MS (ES+) calcd for $\text{C}_{126}\text{H}_{209}\text{N}_{46}\text{O}_{23}$ $[\text{M}+\text{H}]^+$: 2734.7, found: 2735.0; $[\text{M}+\text{K}]^+$: 2773.7, found: 2775.0.

(*His*)₈(*LysPhe*)₄(*LysPro*)₂*Lys^oProNH₂* (**H7**). Starting with 250 mg of TG S RAM ($0.24 \text{ mmol} \cdot \text{g}^{-1}$), dendrimer **H7** was obtained as a white foamy solid after cleavage from the resin and preparative RP-HPLC purification (15.8 mg, 7%). Anal. RP-HPLC: $t_{\text{R}} = 7.7$ min. MS (ES+) calcd for $\text{C}_{141}\text{H}_{202}\text{N}_{47}\text{O}_{22}$ $[\text{M}+\text{H}]^+$: 2905.6, found: 2906.0; $[\text{M}+\text{K}]^+$: 2943.6, found: 2946.0.

(*His*)₈(*Lys β Ala*)₄(*LysThr*)₂*Lys^oProNH₂* (**H8**). Starting with 250 mg of TG S RAM ($0.24 \text{ mmol} \cdot \text{g}^{-1}$), dendrimer **H8** was obtained as a white foamy solid after cleavage from the resin and preparative RP-HPLC purification (3.3 mg, 2%). Anal. RP-HPLC: $t_{\text{R}} = 7.0$ min. MS (ES+) calcd for $\text{C}_{115}\text{H}_{186}\text{N}_{47}\text{O}_{24}$ $[\text{M}+\text{H}]^+$: 2609.5, found: 2609.0.

(*His*)₈(*Lys*)₄(*LysGly*)₂*LysLysOH* (**H9**). Starting with 300 mg of TGA ($0.26 \text{ mmol} \cdot \text{g}^{-1}$), dendrimer **H9** was obtained as a white foamy solid after cleavage from the resin and preparative RP-HPLC purification (73.8 mg, 29%). Anal. RP-UPLC: $t_{\text{R}} = 1.212$ min (A/D = 100/0 to 0/100 in 2.2 min, 100 D in 1.3 min, A/D = 0/100 to 100/0 in 0.5 min, 100 A in 1 min $1.2 \text{ mL} \cdot \text{min}^{-1}$). MS (ES+) calcd for $\text{C}_{100}\text{H}_{161}\text{N}_{42}\text{O}_{19}$ $[\text{M}+\text{H}]^+$: 2254.3, found: 2254.0; $[\text{M}+\text{Na}]^+$: 2276.3, found: 2277.0; $[\text{M}+\text{K}]^+$: 2293.3, found: 2295.0.

(*Phe*)₈(*LysHis*)₄(*LysHis*)₂*LysLysOH* (**H10**). Starting with 300 mg of TGA ($0.26 \text{ mmol} \cdot \text{g}^{-1}$), dendrimer **H10** was obtained as a white foamy solid after cleavage from the resin and preparative RP-HPLC purification (64.0 mg, 21%). Anal. RP-HPLC:

$t_R = 7.3$ min ($A/D = 100/0$ to $0/100$ in 7 min, 100 D in 3 min, 1.4 mL·min⁻¹). MS (ES+) calcd for C₁₅₆H₂₁₃N₄₂O₂₃ [M+H]⁺: 3042.7, found: 3043.0; [M+K]⁺: 3082.7, found: 3083.0.

(AcSer)₈(Lys)₄(LysTyr)₂Lys^αProNH₂ (**AcN1**). Starting with 250 mg of TG S RAM (0.24 mmol·g⁻¹), dendrimer **AcN1** was obtained as a white foamy solid after cleavage from the resin and preparative RP-HPLC purification (7.9 mg, 5%). Anal. RP-HPLC: $t_R = 7.7$ min. MS (ES+) calcd for C₁₀₅H₁₇₀N₂₇O₃₆ [M+H]⁺: 2385.2, found: 2385.0; [M+Na]⁺: 2407.2, found: 2407.0; [M+K]⁺: 2423.2, found: 2423.0.

(AcPhe)₈(LysArg)₄(LysHis)₂LysLysOH (**AcN2**). Starting with 300 mg of TGA (0.26 mmol·g⁻¹), dendrimer **AcN2** was obtained as a white foamy solid after cleavage from the resin and preparative RP-HPLC purification (28.8 mg, 9%). Anal. RP-UPLC: $t_R = 1.777$ min ($A/D = 100/0$ to $0/100$ in 2.2 min, 100 D in 1.3 min, $A/D = 0/100$ to $100/0$ in 0.5 min, 100 A in 1 min, 1.2 mL·min⁻¹). MS (ES+) calcd for C₁₇₂H₂₄₉N₄₆O₃₁ [M+H]⁺: 3454.9, found: 3455.0; [M+TFA]⁺: 3567.9, found: 3570.0.

(Ala)₈(LysPro)₄(LysGln)₂Lys^αProNH₂ (**N3**). Starting with 250 mg of TG S RAM (0.24 mmol·g⁻¹), dendrimer **N3** was obtained as a white foamy solid after cleavage from the resin and preparative RP-HPLC purification (1.3 mg, 1%). Anal. RP-HPLC: $t_R = 6.6$ min. MS (ES+) calcd for C₁₀₁H₁₈₀N₃₃O₂₄ [M+H]⁺: 2239.4, found: 2239.0; [M+K]⁺: 2277.4, found: 2278.0.

(Ser)₈(LysVal)₄(LysHis)₂LysLysOH (**N4**). Starting with 300 mg of TGA (0.26 mmol·g⁻¹), dendrimer **N4** was obtained as a white foamy solid after cleavage from the resin and preparative RP-HPLC purification (46.5 mg, 22%). Anal. RP-HPLC: $t_R = 6.9$ min. MS (ES+) calcd for C₁₀₄H₁₈₉N₃₄O₃₁ [M+H]⁺: 2410.4, found: 2410.0; [M+Na]⁺: 2432.4, found: 2432.0; [M+K]⁺: 2449.4, found: 2450.0.

Isobutyryl Fluorescein (2). Fluorescein (500 mg, 1.5 mmol) was added to a suspension of 43 mg NaH (1.2 equiv.) and 50 mg ZnCl₂ (0.25 equiv.) in 10 mL of anhydrous DMF in the presence of molecular sieves. Isobutyryl chloride (160 μL, 1 equiv.) in 0.25 mL of dry DMF was added dropwise to the first solution for 5 min. After 18 h the reaction mixture was poured into 100 mL of ethyl acetate, washed with brine (3 × 100 mL), and dried over Na₂SO₄. The solvents were removed, and the crude was purified with flash chromatography (hexane-ethyl acetate 6:4) to yield 233 mg of pure **2**. Yield 38%. ¹H NMR (CDCl₃, 400 MHz) δ = 8.04–7.99 (m, 1H), 7.69–7.56 (m, 2H), 7.11–7.04 (m, 2H), 6.83–6.72 (m, 2H), 6.69 (d, $J = 2.3$ Hz, 1H), 6.55 (dt, $J = 8.7, 5.5$ Hz, 2H), 2.83 (dt, $J = 17.7, 7.0$ Hz, 1H), 1.33 (d, $J = 1.2$ Hz, 3H), 1.31 (d, $J = 1.2$ Hz, 3H). ¹³C NMR (CDCl₃, 400 MHz) δ = 175.8, 170.1, 158.7, 152.9, 152.6, 152.5, 152.2, 135.4, 130.1, 129.5, 126.7, 125.5, 124.4, 117.6, 116.8, 113.1, 110.9, 110.6, 103.4, 34.5, 19.1. Hrms (ESI+) Calculated for C₂₄H₁₉O₆ [M+H]⁺, 403.1181, observed 403.1176, [M+Na]⁺ 426.39, observed 425.10.

Kinetic Measurements. Peptides solutions were used as 11.5 μM (His10-His14), 15 μM (all dendrimers), 22.5 μM (His8-His9), 45 μM (His5-His7) for substrate **1**, 11.5 μM (His11-His15, P25), 15 μM (all dendrimers), 22.5 μM (His8-His10), 45 μM (His5-His7), 60 μM (His1-His4) for substrate **2**, 11.5 μM (His15, P25), 15 μM (all dendrimers, His7-His14), 22.5 μM (His8-His10), 45 μM (His5-His6), 60 μM (His1-His4) for substrate **4**, as freshly prepared solutions in mQ-H₂O. Phosphate buffer 60 mM pH 7.6, citric acid-sodium citrate buffer 15 mM at

pH 5.5 were used as buffer, and the pH was adjusted to the desired value with HCl 1.0 M or NaOH 1.0 M using a Metrohm 692 pH/ion meter. Eight final substrate concentrations ranging from 5.8 to 100 μM in MeCN/buffer (17:83) for substrate **1**, 5.8 to 100 μM in MeCN/buffer (11:89) for substrate **2**, 58.5 to 1000 μM in buffer for substrate **4** (dilutions by 2/3) were measured for each Michaelis–Menten plot. Eight solutions of fluorescein sodium salt (FNa) ranging from 0 to 10 μM in MeCN/buffer (17:83) for substrate **1**, in MeCN/buffer (11:89) for substrate **2**, 8-hydroxypyrene-1,3,6-trisulfonic acid, trisodium salt (HPTS) from 0 to 100 μM in buffer for substrate **4** were used for the calibration curve. A 40 μL portion of dendrimer was mixed with 40 μL of buffer and 40 μL of substrate in a 96-well half area flat bottom plate (190 μL). The formation of FNa or HPTS was followed by fluorescence emission using absorbance filter 450/50 and emission filter 530/25. The calibration curve and the blank (40 μL substrate, 40 μL buffer and 40 μL mQ-H₂O) were recorded for every experiment in parallel. Prior to every experiment the gain was adjusted to a signal of ~50000 for the maximum concentration of FNa (for **1** and **2**) or HPTS (for **4**) well. The temperature inside the instrument was set to 34 °C. Kinetic experiments were followed for typically 1–3 h, and the data points were measured every 90 s. Fluorescence data were converted into product concentration by means of the calibration curve. Initial reaction rates were calculated from the steepest linear part observed in the curve that gives fluorescence versus time.

Kinetic Parameters k_{cat} and K_M . V_{cat} is the apparent rate in the presence of dendrimer catalyst, and V_{uncat} is the rate in buffer alone. The observed rate enhancement is defined as V_{net}/V_{uncat} with $V_{net} = V_{cat} - V_{uncat}$. Michaelis–Menten parameters k_{cat} (rate constant) and K_M (Michaelis constant) were obtained by fitting the data to the Michaelis–Menten model. With substrate **2** where saturation was not reached no determination of k_{cat} and K_M was possible and only the specificity constant k_{cat}/K_M is given. The rate constant k_{uncat} without catalyst was calculated from the slope of the linear curve that gives V_{uncat} (as product concentration per time) versus substrate concentration [S].

k_2 (4-Methylimidazole). The solutions of 4-methylimidazole were prepared by serial dilution from a stock solution (3 mM) adjusted to the desired pH value using HCl 1 M. The reaction rate with 4-methylimidazole was obtained under the same conditions as described above. The final concentrations in the plate were 0, 88, 132, 198, 296, 444, 667 and 1000 μM of 4-MeIm, 200 μM of substrate, and 60 mM of buffer. The second order rate constants k_2 were calculated from linear regression of the experimentally measured pseudo first order rate constants k_2' as a function of 4-methylimidazole concentrations. The second order rate constants k_2 is given by $k_2 = k_2'/[S]$.

■ ASSOCIATED CONTENT

Supporting Information. Tables S1–S3, catalytic dendrimers (HPLC traces, MS and kinetics spectra). This material is available free of charge via the Internet at <http://pubs.acs.org>.

■ AUTHOR INFORMATION

Corresponding Author

*Fax: + 41 31 631 80 57. E-mail: jean-louis.reymond@ioc.unibe.ch.

ACKNOWLEDGMENT

This work was supported financially by the University of Berne, the Swiss National Science Foundation, and the Marie Curie Training Network IBAAC.

REFERENCES

- (1) (a) Newkome, G. R.; Moorefield, C. N.; Vögtle, F. *Dendritic Molecules: Concepts, Synthesis, Applications*; VCH: Weinheim, 2001; (b) Kofoed, J.; Reymond, J.-L. Dendrimers as artificial enzymes. *Curr. Opin. Chem. Biol.* **2005**, *9*, 656–664. (c) Lee, C. C.; MacKay, J. A.; Fréchet, J. M. J.; Szoka, F. C. Designing dendrimers for biological applications. *Nat. Biotechnol.* **2005**, *23*, 1517–1526. (d) Astruc, D.; Boisselier, E.; Ornelas, C. Dendrimers Designed for Functions: From Physical, Photophysical, and Supramolecular Properties to Applications in Sensing, Catalysis, Molecular Electronics, Photonics, and Nanomedicine. *Chem. Rev.* **2010**, *110*, 1857–1959.
- (2) (a) Kreiter, R.; Kleij, A. W.; Klein Gebbink, R. J. M.; van Koten, G. Dendritic Catalysts. In *Topics in Current Chemistry*; Springer-Verlag: Berlin, 2001; Vol. 217, pp 163–199. (b) Garcia-Martinez, J. C.; Lezutekong, R.; Crooks, R. M. Dendrimer-Encapsulated Pd Nanoparticles as Aqueous, Room-Temperature Catalysts for the Stille Reaction. *J. Am. Chem. Soc.* **2005**, *127*, 5097–5103. (c) Lu, S. M.; Alper, H. Intramolecular Carbonylation Reactions with Recyclable Palladium-Complexed Dendrimers on Silica: Synthesis of Oxygen, Nitrogen, or Sulfur-Containing Medium Ring Fused Heterocycles. *J. Am. Chem. Soc.* **2005**, *127*, 14776–14784. (d) Tang, W.-J.; Yang, N.-F.; Yi, B.; Deng, G.-J.; Huang, Y.-Y.; Fan, Q.-H. Phase selectively soluble dendrimer-bound osmium complex: a highly effective and easily recyclable catalyst for olefin dihydroxylation. *Chem. Commun.* **2004**, 1378–1379. (e) Liang, C. O.; Helms, B.; Craig, J.; Fréchet, J. M. J. Dendronized cyclopolymers with a radial gradient of polarity and their use to catalyze a difficult esterification. *Chem. Commun.* **2003**, 2524–2525.
- (3) (a) Esposito, A.; Delort, E.; Lagnoux, D.; Djojo, F.; Reymond, J.-L. Catalytic Peptide Dendrimers. *Angew. Chem., Int. Ed.* **2003**, *42*, 1381–1383. (b) Douat-Casassus, C.; Darbre, T.; Reymond, J.-L. Selective Catalysis with Peptide Dendrimers. *J. Am. Chem. Soc.* **2004**, *126*, 7817–7826. (c) Delort, E.; Darbre, T.; Reymond, J.-L. A Strong Positive Dendritic Effect in a Peptide Dendrimer-Catalyzed Ester Hydrolysis Reaction. *J. Am. Chem. Soc.* **2004**, *126*, 15642–15643. (d) Delort, E.; Nguyen-Trung, N.-Q.; Darbre, T.; Reymond, J.-L. Synthesis and Activity of Histidine Containing Catalytic Peptide Dendrimers. *J. Org. Chem.* **2006**, *71*, 4468–4480. (e) Darbre, T.; Reymond, J.-L. Peptide Dendrimers as Artificial Enzymes, Receptors, and Drug-Delivery Agents. *Acc. Chem. Res.* **2006**, *39*, 925–934.
- (4) (a) Berkessel, A.; Herault, D. A. Discovery of peptide-zirconium complexes that mediate phosphate hydrolysis by batch screening of a combinatorial undecapeptide library. *Angew. Chem., Int. Ed.* **1999**, *38*, 102–105. (b) Copeland, G. T.; Miller, S. J. Selection of enantioselective acyl transfer catalysts from a pooled peptide library through a fluorescence-based activity assay. An approach to kinetic resolution of secondary alcohols of broad structural scope. *J. Am. Chem. Soc.* **2001**, *113*, 6496–6502.
- (5) (a) Krattiger, P.; McCarthy, C.; Pfaltz, A.; Wennemers, H. Catalyst-Substrate Coimmobilization. A Strategy for Catalysts Discovery in Split-and-Mix Libraries. *Angew. Chem., Int. Ed.* **2003**, *42*, 1722–1724. (b) Bauke-Albada, H.; Liskamp, R. M. J. TAC-Scaffolded Tripeptides as Artificial Hydrolytic Receptors. A Combinatorial Approach Toward Esterase Mimics. *J. Comb. Chem.* **2008**, *10*, 814–824.
- (6) (a) Furka, A.; Sebestyén, F.; Asgedom, M.; Dibó, G. General method for rapid synthesis of multicomponent peptide mixtures. *Int. J. Pept. Protein Res.* **1991**, *37*, 487–493. (b) Lam, K. S.; Salmon, S. E.; Hersh, E. M.; Hruby, V. J.; Kazmierski, W. M.; Knapp, R. J. A new type of synthetic peptide library for identifying ligand-binding activity. *Nature* **1991**, *354*, 82–84. (c) Houghten, R. A.; Pinilla, C.; Blondelle, S. E.; Appel, J. R.; Dooley, C. T.; Cuervo, J. H. Generation and use of synthetic peptide combinatorial libraries for basic research and drug discovery. *Nature* **1991**, *354*, 84–86. (d) Lam, K. S.; Lebl, M.; Krchnak, V. The “One-Bead-One-Compound” Combinatorial Library Method. *Chem. Rev.* **1997**, *97*, 411–448.
- (7) (a) Clouet, A.; Darbre, T.; Reymond, J.-L. A Combinatorial Approach to Catalytic Peptide Dendrimers. *Angew. Chem., Int. Ed.* **2004**, *43*, 4612–4615. (b) Maillard, N.; Clouet, A.; Darbre, T.; Reymond, J.-L. Combinatorial Libraries of Peptide Dendrimers: Design, Synthesis, On-Bead High-Throughput Screening, Bead Decoding and Characterization. *Nat. Protoc.* **2009**, *4*, 132–142. (c) Javor, S.; Delort, E.; Darbre, T.; Reymond, J.-L. Peptide Dendrimer Enzyme Model with a Single Catalytic Site at the Core. *J. Am. Chem. Soc.* **2007**, *129*, 13238–13246.
- (8) Kofoed, J.; Darbre, T.; Reymond, J.-L. Artificial aldolases from peptide dendrimer combinatorial libraries. *Org. Biomol. Chem.* **2006**, 3268–3281.
- (9) Uhlich, N. A.; Sommer, P.; Bühr, C.; Schürch, S.; Reymond, J.-L.; Darbre, T. Remote Control of Bipyridine-Metal Coordination with a Peptide Dendrimer. *Chem. Commun.* **2009**, 6237–6239.
- (10) (a) Sommer, P.; Uhlich, N. A.; Reymond, J.-L.; Darbre, T. A Peptide Dendrimer Model for Vitamin B₁₂ Transport Proteins. *ChemBioChem* **2008**, *9*, 689–693. (b) Uhlich, N.; Natalello, A.; Kadam, R. U.; Doglia, S. M.; Reymond, J.-L.; Darbre, T. Structure and Binding of Peptide Dendrimer Ligands to Vitamin B₁₂. *ChemBioChem* **2010**, *11*, 358–365.
- (11) (a) Kolomiets, E.; Johansson, E. M. V.; Renaudet, O.; Darbre, T.; Reymond, J.-L. Neoglycopeptide Dendrimer Libraries as a Source of Lectin Binding Ligands. *Org. Lett.* **2007**, *9*, 1465–1468. (b) Kolomiets, E.; Swiderska, M. A.; Kadam, R. U.; Johansson, E. M. V.; Jaeger, K.-E.; Darbre, T.; Reymond, J.-L. Glycopeptide dendrimers with high affinity for the fucose binding lectin PA-III from *Pseudomonas Aeruginosa*. *ChemMedChem* **2009**, *4*, 562–569.
- (12) Maillard, N.; Darbre, T.; Reymond, J.-L. Identification of Catalytic Peptide Dendrimers by “Off-Bead” in Silica High-Throughput Screening of Combinatorial Libraries. *J. Comb. Chem.* **2009**, *11*, 667–675.
- (13) (a) Kofoed, J.; Nielsen, J.; Reymond, J.-L. Discovery of New Peptide-based Catalysts for the Direct Asymmetric Aldol Reaction. *Bioorg. Med. Chem. Lett.* **2003**, *13*, 2445–2447. (b) Martin, H. J.; List, B. Mining sequence space for asymmetric aminocatalysis. N-Terminal prolylpeptides efficiently catalyze enantioselective aldol and Michael reactions. *Synlett* **2003**, 1901–1902. (c) Tang, Z.; Yang, Z.-H.; Cun, L.-F.; Gong, L.-Z.; Mi, A.-Q.; Jiang, Y.-Z. Small Peptides Catalyze Highly Enantioselective Direct Aldol Reactions of Aldehydes with Hydroxyacetone: Unprecedented Regiocontrol in Aqueous Media. *Org. Lett.* **2004**, *6*, 2285–2287. (d) Notz, W.; List, B. Catalytic Asymmetric Synthesis of anti-1,2-Diols. *J. Am. Chem. Soc.* **2000**, *122*, 7386–7387. (e) Shi, L.-X.; Sun, Q.; Ge, Krattiger, P.; Kovasy, R.; Revell, J. D.; Ivan, S.; Wennemers, H. Increased structural complexity leads to higher activity: peptides as efficient and versatile catalysts for asymmetric aldol reactions. *Org. Lett.* **2005**, *7*, 1101–1103.
- (14) (a) Kofoed, J.; Reymond, J. L. A general method for designing combinatorial peptide libraries decodable by amino acid analysis. *J. Comb. Chem.* **2007**, *9*, 1046–52. (b) Kofoed, J.; Reymond, J. L. Identification of protease substrates by combinatorial profiling on TentaGel beads. *Chem. Commun.* **2007**, 4453–4455. (c) Fluxa, V. S.; Reymond, J. L. On-bead cyclization in a combinatorial library of 15,625 octapeptides. *Bioorg. Med. Chem.* **2009**, *17*, 1018–1025. (d) Dulery, V.; Uhlich, N. A.; Maillard, N.; Fluxa, V. S.; Garcia, J.; Dumy, P.; Renaudet, O.; Reymond, J. L.; Darbre, T. A cyclodecapeptide ligand to vitamin B₁₂. *Org. Biomol. Chem.* **2008**, *6*, 4134–4141. (e) Fluxa, V. S.; Maillard, N.; Page, M. G.; Reymond, J. L. Bead diffusion assay for discovering antimicrobial cyclic peptides. *Chem. Commun.* **2011**, 1434–1436.
- (15) Biswas, R.; Maillard, N.; Kofoed, J.; Reymond, J.-L. Comparing dendritic with linear esterase peptides by screening SPOT arrays for catalysis. *Chem. Commun.* **2010**, 8746–8748.
- (16) Yang, Y.; Babiak, P.; Reymond, J.-L. New Monofunctionalized Fluorescein Derivatives for the Efficient High-Throughput Screening of Lipases and Esterases in Aqueous Media. *Helv. Chim. Acta* **2006**, *89*, 404–415.

(17) Blankemeyer-Menge, B.; Nimtz, M.; Frank, R. An efficient method for anchoring Fmoc-amino acids to hydroxyl-functionalized solid supports. *Tetrahedron Lett.* **1990**, *31*, 1701–1704.

(18) Attardi, M. E.; Falchi, A.; Taddei, M. A sensitive visual test for detection of OH groups on resin. *Tetrahedron Lett.* **2000**, *41*, 7395–7399.

(19) (a) Hancock, W. S.; Battersby, J. E. A new micro-test for the detection of incomplete coupling reactions in solid-phase peptide synthesis using 2,4,6-trinitrobenzenesulfonic acid. *Anal. Biochem.* **1976**, *71*, 260–264. (b) Vojkovsky, T. Detection of secondary amines on solid phase. *Pept. Res.* **1995**, *8*, 236–237.